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In order to use Sutori, you must enable Javascript in your browser. You can find instructions on how to do this here. BALANCING CHEMICAL EQUATIONS 1 BALANCING CHEMICAL EQUATIONS Load the simulation Balancing Chemical Equations CRITICAL THINKING QUESTIONS 1. Explore the Balancing Chemical Equations simulation. a) What are the different ways that the simulation indicates when an equation is balanced? A face smiles, scales and 2 arrows turn yellow 2. For each balanced reaction, indicate the total number of molecules in the table below. Total Number of Molecules 3. Is the number of total molecules on the left side of a balanced equation always equal to the number of total molecules on the right side of the equation? Explain your answer. No. the number of molecules does not necessarily have to be the same. 4. For each balanced reaction, indicate the total number of atoms in the table below. 5. Is the number of total atoms on the left side of a balanced equation always equal to the number of total atoms on the right side of the equation? Yes 6. What is the same on the left and right side of a balanced equation? Explain your answer. The number of atoms. – law of conservation of mass 7. As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations. a.  $2CO_2 \rightarrow 2CO + 1O_2$  b.  $1PCl_5 \rightarrow 1PCl_3 + 1Cl_2$  c.  $1P_4 + 6F_2 \rightarrow 4PF_3$  d.  $1C + 1O_2 \rightarrow 1CO_2$

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